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## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification 6 :</b> C08L 67/02 // (C08L 67/02, 77:12)	<b>A1</b>	<b>(11) International Publication Number:</b> WO 97/28218 <b>(43) International Publication Date:</b> 7 August 1997 (07.08.97)
<b>(21) International Application Number:</b> PCT/US97/01638 <b>(22) International Filing Date:</b> 3 February 1997 (03.02.97) <b>(30) Priority Data:</b> 08/595,460 5 February 1996 (05.02.96) US <b>(71) Applicant:</b> EASTMAN CHEMICAL COMPANY [US/US]; 100 North Eastman Road, Kingsport, TN 37660 (US). <b>(72) Inventors:</b> TURNER, Sam, Richard; 1037 Sussex Drive, Kingsport, TN 37660 (US). NICELY, Vincent, Alvin; 2605 Suffolk Street, Kingsport, TN 37660 (US). <b>(74) Agent:</b> HARDING, Karen, A.; P.O. Box 511, Kingsport, TN 37662-5075 (US).		<b>(81) Designated States:</b> AU, BR, BY, CA, CN, CZ, HU, IL, JP, KR, MX, NO, NZ, PL, RU, SG, SK, TR, UA, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).  <b>Published</b> <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
<b>(54) Title:</b> POLYESTER/POLYESTERAMIDE BLENDS		
<b>(57) Abstract</b> <p>This invention relates to polyester polyesteramide blends comprising: (A) about 95-99.99 weight percent of a polyester which comprises a dicarboxylic acid component comprising repeat units from at least 85 mol % terephthalic acid, naphthalene-2,6-dicarboxylic acid or mixtures of terephthalic acid and naphthalene-2,6-dicarboxylic acid and a diol component comprising repeat units from at least 85 mol % ethylene glycol, said mole percents being based on 100 mol % dicarboxylic acid and at least 100 mol % diol; and (B) about 5-0.01 weight percent of a polyesteramide of the formula: <math>A(N)_X(D)_Y</math>, where A is a dicarboxylic acid selected from aromatic or cycloaliphatic dicarboxylic acids having 8 to 14 carbon atoms or aliphatic dicarboxylic acids having 3 to 24 carbon atoms, N is a diamine having 2 to 24 carbon atoms, D is a diol having 2 to 14 carbon atoms, X is an integer from 0.01 to 0.99 and Y is an integer from 0.99 to 0.01. The blends of the present invention display good barrier properties, low color and acetaldehyde concentration. The blends are useful for a variety of molded articles including containers, films and sheets.</p>		

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POLYESTER/POLYESTERAMIDE BLENDSTechnical Field

This invention relates to a polyester/  
5 polyesteramide blend having excellent gas barrier and  
improved flavor retaining properties and color. Certain  
polyethylene terephthalate/polyesteramide blends of the  
present invention also display reduced acetaldehyde  
concentration which highly improves the flavor and  
10 fragrance retaining properties without producing  
unacceptable haze or color.

Background of the Invention

Polyethylene terephthalate (PET) is widely used for  
15 the production of light weight plastic articles due to  
PET's excellent mechanical properties such as  
formability, creep resistance and biaxial molecular  
orientability. However, during molding or extrusion  
processes, acetaldehyde is formed by thermal  
20 decomposition of the polyester and when the polyester is  
formed into an article, the acetaldehyde in the  
article's walls migrates into the contents of the  
article. Small amounts of acetaldehyde adversely affect  
the flavor retaining property of foods and beverages,  
25 and the fragrance retaining property of foods,  
beverages, cosmetics, and other package contents. For  
these reasons, it is desirable to minimize the migration  
of acetaldehyde into package contents.

Thermoplastic polyesters excellent in gas barrier  
30 properties have been proposed. For example, U.S. Patent  
No. 4,398,017 discloses a copolyester made from  
terephthalic acid and isophthalic acid as the acid  
component, and ethylene glycol and bis(2-hydroxyethoxy)  
benzene as the diol component. However, when such gas  
35 barrier polyester is used as a constituent material of a

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container, permeation of gases such as oxygen and carbon dioxide are controlled, but acetaldehyde migration into the food or beverage is not controlled, and thus, the flavor and fragrance of the content is effected.

5       The use of polyamides to increase the gas barrier properties in polyethylene terephthalate resins is disclosed in U.S. Patent Nos. 4,837,115; 4,052,481; and 4,501,781.

10       U.S. Patent No. 4,837,115 discloses a thermoplastic composition containing polyethylene terephthalate and high molecular weight polyamides which act to reduce the residual acetaldehyde contained in the polyester. U.S. Patent No. 4,837,115 teaches that terminal amino groups are a required component for the polyamides to be  
15       effective acetaldehyde scavengers. U.S. Patent No. 4,837,115 states that the molecular weight of the polyamide is not critical so far as the polyamide has a film-forming property. Such polyamides, therefore, must have high enough molecular weights to form a film. It  
20       is well known in the art that polyamides having molecular weights of at least 12,000 are necessary to form a film.

25       U.S. Patent No. 4,052,481, discloses a resin composition containing an aromatic copolyester, a polyamide and a polyalkylene phenylene ester or a polyalkylene phenylene ester ether. The aromatic copolyester contains terephthalic acid, isophthalic acid, and a bisphenol.

30       U.S. Patent No. 4,501,781 discloses a mixture containing 70 to 95 weight percent of a polyethylene terephthalate resin and 5 to 30 weight percent of a xylylene group-containing polyamide resin. U.S. Patent No. 4,501,781 does not mention acetaldehyde, however, it does state that in order to mold a container that has  
35       high gas barrier properties, the resin mixture material

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should have as close to 30% by weight of the xylylene group-containing polyamide resin to PET as possible. Furthermore, it states that the use of 5 to 10% by weight of the xylylene group-containing polyamide resin to PET will result in a container that does not have high gas barrier properties.

Japan Kokai Patent No. 5H0 64[1989]-24849 discloses blends of aliphatic polyamides in polyethylene terephthalate which provide reduced acetaldehyde after melt processing. Such blends will decrease the desired transparency of containers made from them.

The above mentioned patents are deficient because high molecular weight polyamides do not provide adequate reduction in residual acetaldehyde without imparting haze to the polyester. If small amounts of the high molecular weight polyamides are used in such patents, an acceptable level of haze can be achieved, however, residual acetaldehyde is very large. On the other hand, if larger amounts of the high molecular weight polyamides are used, residual acetaldehyde can be reduced but only at the expense of haze.

U.S. 5,258,233 discloses the use of low molecular weight polyamides to reduce acetaldehyde levels in PET based polyester without the formation of undesirable amounts of haze. However, it has been observed that the use of these low molecular weight polyamides can lead to undesirable yellow color.

We have unexpectedly discovered that polyesteramides, which have no terminal amino functionality, act as very efficient acetaldehyde reducers with the advantage of significantly less color formation. This is particularly desirable because high levels of color in food packaging is unacceptable. Moreover, in contrast to the polyamides, we have

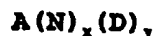
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observed that high molecular weight polyesteramides are effective for acetaldehyde reduction.

#### Summary of the Invention

5       The present invention provides semi-crystalline polyester compositions having improved flavor retaining properties, comprising:

- 10       I. 95.0 to 99.99 weight percent of a polyester which comprises
- (1) a dicarboxylic acid component comprising repeat units of terephthalic acid, naphthalene-2,6-dicarboxylic acid or mixtures of terephthalic acid and naphthalene-
- 15       2,6-dicarboxylic acid,
- (2) a diol component comprising repeat units from at least 85 mole percent ethylene glycol, based on 100 mole percent dicarboxylic acid and 100 mole percent diol; and
- 20       II. 5-0.01 weight percent of a polyesteramide of the formula



25       where A is at least one dicarboxylic acid selected from aromatic or cycloaliphatic dicarboxylic acids having 8 to 14 carbon atoms or aliphatic dicarboxylic acids having 3 to 24 carbon atoms, N is at least one diamine having 2 to 24 carbon

30       atoms, D is at least one diol having 2 to 14 carbon atoms, X is an integer from 0.01 to 0.99 and Y is an integer from 0.99 to 0.01.

The combined weights of the polyester in the blend and the polyesteramide in the blend total 100%. The

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preferred weight percent of the polyester is 98.0 to 99.95 and the polyesteramide 2.0 to 0.05.

#### Description of the Invention

5       The polyester, component (I), of the present invention is a polyethylene terephthalate (PET) or polyethylene naphthalate (PEN) resin. Copolyesters and blends of PET and PEN can also be used. The polyethylene terephthalate resin contains repeat units  
10       from at least 85 mole percent terephthalic acid and at least 85 mole percent ethylene glycol, while the PEN resin contains repeat units from at least 85 mole percent 2,6-naphthalene-dicarboxylic acid and at least  
15       85% ethylene glycol, based on 100 mole percent dicarboxylic acid and 100 mole percent diol.

      The dicarboxylic acid component of the polyester may optionally be modified with up to 15 mole percent of one or more different dicarboxylic acids other than terephthalic acid or suitable synthetic equivalents such  
20       as dimethyl terephthalate. Such additional dicarboxylic acids include aromatic dicarboxylic acids preferably having 8 to 14 carbon atoms, aliphatic dicarboxylic acids preferably having 4 to 12 carbon atoms, or cycloaliphatic dicarboxylic acids preferably having 8 to  
25       12 carbon atoms. Examples of dicarboxylic acids to be included with terephthalic acid are: phthalic acid, isophthalic acid, naphthalene-2,6- dicarboxylic acid, cyclohexanedicarboxylic acid, cyclohexanediacetic acid, diphenyl-4,4'-dicarboxylic acid, succinic acid, glutaric  
30       acid, adipic acid, azelaic acid, sebacic acid, and the like. Examples of dicarboxylic acids to be included with naphthalene- 2,6-dicarboxylic acid are phthalic acid, isophthalic acid, naphthalene-2,6-dicarboxylic acid, cyclohexanedicarboxylic acid, cyclohexanediacetic acid,  
35       diphenyl-4,4'-dicarboxylic acid, succinic acid, glutaric

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acid, adipic acid, azelaic acid, sebacic acid, and the like. Polyesters may be prepared from two or more of the above dicarboxylic acids.

5 It should be understood that use of the corresponding acid anhydrides, esters, and acid chlorides of these acids is included in the term "dicarboxylic acid".

In addition, the polyester, component (I), may optionally be modified with up to 15 mole percent, of one or more different diols other than ethylene glycol. Such additional diols include cycloaliphatic diols preferably having 6 to 20 carbon atoms or aliphatic diols preferably having 3 to 20 carbon atoms. Examples of such diols to be included with ethylene glycol are:

15 diethylene glycol, triethylene glycol, 1,4-cyclohexanedimethanol, propane-1,3-diol, butane-1,2-diol, pentane-1,5-diol, hexane-1,6-diol, 3-methylpentanediol-(2,4), 2-methylpentanediol-(1,4), 2,2,4-trimethylpentane-diol-(1,3), 2-ethylhexanediol-(1,3), 2,2-diethylpropane-diol-(1,3), hexanediol-(1,3), 20 1,4-di-(hydroxyethoxy)-benzene, 2,2-bis-(4-hydroxy-cyclohexyl)-propane, 2,4-dihydroxy-1,1,3,3-tetramethyl-cyclobutane, 2,2-bis-(3-hydroxyethoxyphenyl)-propane, and 2,2-bis-(4-hydroxypropoxyphenyl)-propane.

25 Polyesters may be prepared from two or more of the above diols.

The polyethylene terephthalate resin may also contain small amounts of trifunctional or tetrafunctional comonomers such as trimellitic anhydride, trimethylolpropane, pyromellitic dianhydride, 30 pentaerythritol, and other polyester forming polyacids or polyols generally known in the art.

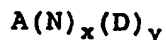
Polyesters comprising substantially only terephthalic acid or dimethyl terephthalate and ethylene 35 glycol residues are preferred in the case where the

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blends of the present invention are used in making thermoformed crystallized PET articles.

Polyethylene terephthalate based polyesters of the present invention can be prepared by conventional polycondensation procedures well-known in the art. Such processes include direct condensation of the dicarboxylic acid(s) with the diol(s) or by ester interchange using a dialkyl dicarboxylate. For example, a dialkyl terephthalate such as dimethyl terephthalate is ester interchanged with the diol(s) at elevated temperatures in the presence of a catalyst. The polyesters may also be subjected to solid state polymerization methods.

The second component of the present invention is a polyesteramide of the general formula:



where A is at least one dicarboxylic acid selected from aromatic or cycloaliphatic dicarboxylic acids having 8 to 14 carbon atoms or aliphatic dicarboxylic acids having 3 to 24 carbon atoms, N is at least one diamine having 2 to 24 carbon atoms, D is at least one diol having 2 to 14 carbon atoms, X is an integer from 0.01 to 0.99 and Y is an integer from 0.99 to 0.01. The most preferred range is for X to vary from 0.25 to 0.74 and Y to range from 0.74 to 0.25. The sum of X and Y is equal to 1.0.

A in this formula represents any of the dicarboxylic acids normally used in polyester and polyamide preparations. It is obvious to one skilled in the art that acid anhydrides, esters, and acid chlorides of these acids can be used to prepare these polyesteramides and are included in the term "dicarboxylic acid."

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Suitable dicarboxylic acids are selected from the group consisting of aromatic dicarboxylic acids having 8 to 16 carbon atoms, aliphatic diacidic acids having 3 to 12 carbon atoms and cycloaliphatic dicarboxylic acids having 8 to 12 carbon atoms.

The dicarboxylic acid component (A) of the polyesteramide may consist of one or more different dicarboxylic acids. Preferably the dicarboxylic acid is terephthalic acid, isophthalic acid, 1,4-cyclohexanedicarboxylic acid, 1,3-cyclohexanedicarboxylic acid, 2,6-naphthalene dicarboxylic acid, succinic acid, adipic acid, azelaic acid, sebacic acid, and the like.

The diol component (D) of the polyesteramide may consist of one or more different diols. D may be represented by, but not limited to, ethylene glycol, diethylene glycol, triethylene glycol, propane diol, 1,4-cyclohexanediol, 1,4-butanediol, 2,2-bis-(4-hydroxycyclohexane)-propane, 1,3-(2,2-dimethyl)propane diol, 1,4-cyclohexanedimethanol, 1,4-di-(hydroxyethoxy) benzene, hydroquinone, bisphenol A or any combination of these diols.

The diamine component (N) of the polyesteramide may consist of one or more different diamines. N may be represented by, but not limited to, ethylene diamine, propane diamine, butane diamine, hexane diamine, 1,4-cyclohexanediamine, 1,3-cyclohexane diamine, 1,4-phenylene diamine, methylene dianiline, 1,3-xylylene diamine, 1,3-(2,2-dimethyl)propane diamine, 1,4 and 1,3-cyclohexanebismethylamine, 1,4-xylylene diamine, bis(p-aminocyclohexyl)methane or any combination of these diamines.

By correctly balancing the stoichiometry the polyesteramide may be prepared using amino alcohols. These may be used without other diamines or diols or

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used in combination with the above described diamines and diols. These amino alcohols can be represented, but not limited to ethanol amine, 4-aminomethylcyclohexanemethanol, and 1-amino-3-hydroxy-2,2-dimethyl propane.

5       The diamine portion of the general structure shown above can also be incorporated via a reactive intermediate prepared by preforming a "monomeric" diamide. For example N,N'-p-carbomethoxybenzoyl-hexamethylene diamine can be used to prepare  
10       polyesteramides as described in G. Manzini, et. al, Eur. Polym J. 9, 941 (1973).

It is not necessary that the structure of the polyesteramide be limited to linear polymers. The use of multifunctional branching amines, carboxylic acids,  
15       or polyols will result in effective acetaldehyde reducing compositions as long as the resulting branched polymer can be effectively dispersed in the polyester. Thus polyols such as 1,1,1-trimethylolpropane and pentaerythritol are useful as well as multifunctional  
20       amines such as tris(2-aminoethyl)amine and multifunctional carboxylic acids such as 1,3,5-benzene tricarboxylic acid. Hyperbranching units such as 3,5-diaminobenzoic acid can also be used to build effective acetaldehyde reducing structures.

25       The polyesteramides of the present invention do not show a significant molecular weight difference in their efficacy of reducing acetaldehyde in polyesters. This is in contrast to the performance of the polyamides described in U.S. 5,258,233 and 5,340,884. For example,  
30       polyesteramides ranging in molecular weight from 500-1,000,000 (number average) show excellent acetaldehyde reducing characteristics. It is obvious to those skilled in the art of blending polymers that solid polyesteramides ranging in number average molecular

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weight of 1,000 to 100,000 are preferred for ease of blend processing.

5 The polyesteramides can be prepared by any of the known techniques for polyesteramide synthesis. Acid chlorides can be reacted with mixtures of diamines and diols, low molecular weight dicarboxylic acid terminated polyesters can be reacted with diisocyanates, and conventional melt phase step-growth condensations of diacids, diamines, and diols or diesters, diols, and 10 diamines can be employed. Aminolysis of polyesters with diamines, as described in U.S. Patent 4,606,449, can be used to prepare polyesteramides. Direct melt condensation of dicarboxylic acids, diamines, and diols is a preferred synthetic route.

15 The composition of the polyesteramide is controlled by the molar ratio of the diamine, diol, and dicarboxylic acid. This can be adjusted to control molecular weight or control the end group functionality. The order of addition of the reactants can also be used 20 to change the structure of the polyesteramide. For example, an initial charge of ester and alcohol followed by addition of the diamine to the prepolymer can yield a "blocky" structure.

25 The polymers of this invention can be prepared in the presence or absence of a polycondensation catalyst. In the event that catalysts are used conventional polyester catalysts including, but not limited to, cobalt acetate, titanium isopropoxide, manganese acetate, antimony oxide, dibutyl tin diacetate.

30 The process for preparing the polyester/polyester-amide blends of the present invention involves preparing the polyester and polyesteramide, respectively, by processes as mentioned previously. The polyester and polyesteramide are dried in an atmosphere of dried air 35 or dried nitrogen, or under reduced pressure. The

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polyester and polyesteramide are mixed and subsequently melt compounded, for example, in a single or twin screw extruder. Melt temperatures must be at least as high as the melting point of the polyester and are typically in the range of 260–310°C. Preferably, the melt compounding temperature is maintained as low as possible within said range. After completion of the melt compounding, the melt may be extruded in strand form and recovered according to the usual way such as cutting or it may be transferred to a machine suitable to form it directly into useful articles. Instead of melt compounding, the polyester and polyesteramide may be dry-blended and heat-molded or draw-formed into plastic articles.

The polyesteramide can be added in the late stages of polyester manufacture. For example, the polyesteramide can be blended with the molten polyester as it is removed from the polycondensation reactor, before it is pelletized. This method, however, is not desirable if the polyester/polyesteramide blend will be subjected to solid state polymerization since undesirable color and/or haze may develop during extended time at elevated temperatures. The polyesteramide may also be added as part of a polyolefin based nucleator concentrate where clarity is not critical such as in crystallized thermoformed articles.

The blends of this invention serve as excellent starting materials for the production of moldings of all types by extrusion or injection molding. Specific applications include various packaging applications such as thermoformed or injection molded trays, lids and cups; injection stretch blow-molded bottle and multilayer articles. Examples of package contents include, but are not limited to, food, beverages, and cosmetics.

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Many other ingredients can be added to the compositions of the present invention to enhance the performance properties of the blends. For example, crystallization aids, impact modifiers, surface  
5 lubricants, denesting agents, stabilizers, antioxidants, ultraviolet light absorbing agents, metal deactivators, colorants such as titanium dioxide and carbon black, nucleating agents such as polyethylene and  
polypropylene, phosphate stabilizers, fillers, and the  
10 like, can be included herein. All of these additives and the use thereof are well known in the art and do not require extensive discussions. Therefore, only a limited number will be referred to, it being understood that any of these compounds can be used so long as they  
15 do not hinder the present invention from accomplishing its objects.

In applications where a clear, colorless resin is desired, the slight yellow color generated during processing can be masked by addition of a blue dye. The  
20 colorant can be added to either component of the blend during polymerization or added directly to the blend during compounding. If added during blending, the colorant can be added either in pure form or as a concentrate. The amount of a colorant depends on its  
25 absorptivity and the desired color for the particular application. A preferred colorant is  
1-cyano-6-(4-(2-hydroxyethyl)anilino)-3-methyl-  
3H-dibenzo(F,I,J)-isoquinoline-2,7-dione used in an  
amount of from 2 to 15 ppm.

Desirable additives also include impact modifiers  
30 and antioxidants. Examples of typical commercially available impact modifiers well-known in the art and useful in this invention include ethylene/propylene terpolymers, styrene based block copolymers, and various  
35 acrylic core/shell type impact modifiers. The impact

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modifiers may be used in conventional amounts from 0.1 to 25 weight percent of the overall composition and preferably in amounts from 0.1 to 10 weight percent of the composition. Examples of typical commercially available antioxidants useful in this invention include, but are not limited to, hindered phenols, phosphites, diphosphites, polyphosphites, and mixtures thereof. Combinations of aromatic and aliphatic phosphite compounds may also be included.

The materials and testing procedures used for the results shown herein are as follows:

Polyester A is a copolyester consisting of 100 mole percent terephthalic acid, 98 to 99 mole percent ethylene glycol and 1 to 2 mole percent 1,4-cyclohexanedimethanol, having an I.V. of 0.76.

Haze was determined by ASTM D1003. Haze values of greater than 3.0% indicate visible haze.

Inherent viscosity (I.V.) was measured at 25°C using 0.50 grams of polymer per 100 ml of a solvent consisting of 60% by weight phenol and 40% by weight tetrachloroethane.

Number average molecular weights of the polyester were determined by size exclusion chromatography.

The following examples are presented to further illustrate this invention. All parts and percentages in the examples are on a weight basis unless otherwise stated.

#### Examples

Example 1. A polyesteramide based on terephthalic acid (TPA), cyclohexanedimethanol (CHDM), and 50% mol % of cyclohexanedimethanolamine (CHBMA) was prepared by charging a polymerization reactor with TPA (83 g, 0.5 mol), CHDM (108 G, 0.75 mol) and 100 ppm of Ti catalyst

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as titanium isopropoxide. The mixture was placed into a 300°C metal bath and the CMBMA was added in equal portions at 30 minute intervals until a total charge of 35.5 g equivalent to 0.25 mol was added. The  
5 polymerization temperature was raised to 305°C and held for 40 minutes and then brought under vacuum (0.10 Torr) for 80 minutes. A light yellow semicrystalline polyesteramide was obtained. Molecular weight determination by size exclusion chromatography showed  
10  $M_n = 11,060$ ;  $M_w = 44,040$ ; and  $MWD = 3.98$ . Elemental analysis and NMR analysis confirmed the expected composition based on the monomer feed.

Example 2. A polyesteramide based on TPA, CHDM, and 30 mol % m-xylylene diamine (MX) was prepared by charging a  
15 polymerization reactor with TPA (58.1 g, 0.35 mole), CHDM (108 g, 0.75 mol) and 100 ppm of Ti catalyst as titanium isopropoxide. The mixture was placed into a 300°C metal bath and a slurry of TPA and MX (24.9 g,  
20 0.15 mol TPA, 20.4 g, 0.15 mol MX) was added in three separate additions at 30 minute intervals. The polymerization temperature was raised to 305°C and held for 40 minutes and then brought under vacuum (0.10 Torr) for 20 minutes. A yellowish semicrystalline solid was  
25 obtained. Molecular weight determination by size exclusion chromatography showed  $M_n = 10,700$ ;  $M_w = 29,100$ ; and  $MWD = 2.72$ . Elemental analysis and NMR analysis confirmed the expected composition based on the monomer feed.

30  
Example 3. A polyesteramide based on TPA, CHDM, and 50 mol % of MX was prepared according to the procedure of Example 2 with the use of 50 mol % of MX rather than 30 mol%. A yellowish amorphous polyesteramide was  
35 obtained. Molecular weight determination by size

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exclusion chromatography showed  $M_n = 10,100$ ;  $M_w = 66,030$ ; and  $MWD = 6.52$ . Elemental analysis and NMR analysis confirmed the expected composition based on the monomer feed.

5

Example 4. A polyesteramide based on TPA, CHDM, and 30 mol % of 1,3-bis(aminomethyl)cyclohexane (BAC) was prepared according to the procedure of Example 2 with the use of 30 mol % of the BAC in place of the MX. A yellowish amorphous polyesteramide was obtained. Molecular weight determination by size exclusion chromatography showed  $M_n = 17,000$ ;  $M_w = 95,800$ ; and  $MWD = 5.63$ . Elemental analysis and NMR analysis confirmed the expected composition based on the monomer feed.

15

Example 5. A polyesteramide based on TPA, CHDM, and 30 mol % of PACM-20 was prepared by charging a polymerization reactor with TPA (83 g, 0.5 mole), CHDM (108 g, 0.75 mol), bis(p-aminocyclohexyl)methane (PACM-20, 31.5 g, 0.15 mol) and 100 ppm of Ti catalyst as titanium tetraisopropoxide. The mixture was purged with dry nitrogen and placed into a 300°C metal bath. A 90 min. condensation stage at 305°C was followed by 30 min at 305°C. The melt polymerization was held under vacuum of 0.21 Torr for 30 min. On cooling, a light yellow amorphous polymer was obtained with an IV of 0.661,  $M_n = 17,030$ ;  $M_w = 40,800$ ; and  $MWD = 2.39$ . Elemental analysis and NMR analysis confirmed the expected composition based on the monomer feed.

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Example 6. A polyesteramide based on TPA, CHDM, and 50 mol % of PACM-20 was prepared according to the procedure of Example 5 with the use of 50 mol % of the diamine instead of 30 mol% as in Example 5. The resulting

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amorphous light-yellow colored polyesteramide had an IV of 0.855,  $M_n = 14,900$ ;  $M_w = 62,960$ ; and  $MWD = 3.79$ . Elemental analysis and NMR analysis were consistent with the monomer feed composition.

5

Example 7. A polyesteramide based on TPA, CHDM and 30 mol % of hexanediamine (HD) was prepared according to the procedure of Example 5 with the exception that 30 mol % of HD was used instead of the PACM-20. The clear yellowish melt was held under a vacuum of 0.28 Torr for 28 minutes. The molecular weight of the resulting semicrystalline polyesteramide was measured by size exclusion chromatography to give  $M_n = 13,800$ ;  $M_w = 82,970$ ; and  $MWD = 5.99$ . Elemental analysis and NMR analysis were consistent with the monomer feed composition.

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Example 8. A polyesteramide based on TPA, CHDM and 50 mol % of HD was prepared according to the procedure of Example 5 with the exception that 50 mol % of HD was used. The clear slightly yellowish melt was held under a vacuum of 0.18 Torr for 8 minutes. The molecular weight of the resulting semicrystalline polyesteramide was measured by size exclusion chromatography to give  $M_n = 9,600$ ;  $M_w = 38,230$ ; and  $MWD = 3.98$ . Elemental analysis and NMR analysis were consistent with the monomer feed composition.

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Example 9. A polyesteramide was prepared from N,N'-p-carbomethoxybenzoylhexamethylene diamine (33.0 g, 0.075 mol) (prepared according to G. Manzini, et al, Eur. Polym. J. 9, 941 (1973), dimethyl terephthalate (33.98 g, 0.325 mol) and CHDM (39.66 g, 0.275 mol). The polymerization was catalyzed with titanium tetraisopropoxide. It was started at 290°C for 90 min

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and then held for 80 min at 305°C under a vacuum of 0.10 Torr. The very light yellow semicrystalline polymer had an inherent viscosity of 0.725,  $M_n = 14,500$ ;  $M_w = 34,200$ ; and  $MWD = 2.35$ . Elemental analysis was  
5 consistent with the monomer feed composition.

Example 10. A polyesteramide based on 50 mol % TPA, 50 mol % IPA (isophthalic acid), CHDM and 50 mol % HD was prepared according to the procedure of Example 5.  
10 The molecular weight of the resulting polyesteramide was measured by size exclusion chromatography to give  $M_n = 20,300$ ;  $M_w = 50,700$ ;  $MWD = 2.50$ . Elemental analysis and NMR analysis were consistent with the monomer feed composition.

15 Example 11. A polyesteramide based on CHDA (1,4-cyclohexanedicarboxylic acid), CHDM and 50 mol % of HD was prepared according to the procedure of Example 5. The molecular weight of the resulting polyesteramide was  
20 measured by size exclusion chromatography to give  $M_n = 19,990$ ;  $M_w = 51,100$ ;  $MWD = 2.56$ . Elemental analysis and NMR analysis were consistent with the monomer feed composition.

25 Example 12. A polyesteramide based on TPA, CHDM and 50 mol % of 2-methylpentanediamine (MPD) was prepared according to the procedure of Example 5. The molecular weight of the resulting polyesteramide was measured by  
30 size exclusion chromatography to give  $M_n = 20,200$ ;  $M_w = 51,400$ ;  $MWD = 2.54$ . Elemental analysis and NMR analysis were consistent with the monomer feed composition.

Example 13. A polyesteramide based on 50 mol % TPA,  
35 50 mol % CHDA, CHDM and 50 mol % HD was prepared

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according to the procedure of Example 5. The molecular weight of the resulting polyesteramide was measured by size exclusion chromatography to give  $M_n = 19,600$ ;  $M_w = 51,300$ ; MWD 2.62. Elemental analysis and NMR analysis were consistent with the monomer food composition.

Examples 14-27. Compounding of PET with a polyesteramide.

A poly(ethylene terephthalate)/CHDM copolyester (Eastapak® PET copolyester 9921W - 600 g) was dried at 150°C in air and a sample (6.0 g) of polyesteramide from Example 1 was dried at 100°C in a vacuum oven. The samples were physically mixed to give a 1 wt % sample of the polyesteramide in polyester A. The mixture was compounded on a Braebender at 275°C. After compounding, the sample was placed into a freezer to minimize loss of acetaldehyde by diffusion until the acetaldehyde content could be measured. PET/PEA compounds of Examples 16-27 were compounded in a similar fashion.

The color (b\*) and acetaldehyde concentration for the compounded polymers were measured as follows.

Acetaldehyde generation (AA Gen) was determined by the following method. After crystallizing for 30 minutes at 180°C, the pelletized polyester was dried overnight at 120°C in a vacuum oven. A Tinius-Olsen melt indexer was loaded with 5 grams of the polyester and held at the test temperature for five minutes. The molten polyester was extruded into water and stored at a temperature of -40°C until grinding. The sample was ground to 20 mesh or finer and 0.5 grams was placed in a sample tube which was immediately sealed. The sample was analyzed by dynamic headspace gas chromatographic analysis using a Hewlett-Packard 5890 Gas Chromatograph with a Perkin Elmer Automatic Thermal Desorption ATD-50

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as the injection system. Acetaldehyde was desorbed by heating the sample at 150°C for ten minutes. The gas chromatography column had a 30 m by 0.53 mm inside diameter.

5 Color was determined according to ASTM D2244 using a Hunter Color Lab instrument.

Table I compares the efficacy of acetaldehyde reduction and color generation for various polyesteramides with the polyamide of m-xylylene diamine and adipic acid a known effective acetaldehyde reducing additive.

TABLE I. Acetaldehyde Results for  
PET Compounded Polyesteramides

Ex #	PE/PEA	Color (b*)	AA genertn (ppm)	
			275°C	295°C
14	Polyester A (Control)		5.73	16.68
15	Polyester A + 0.5% polyamide (1)	11.49	1.99	8.46
16	Polyester A + 0.5% PEA (2) from Ex. 1	7.47	1.68	4.84
17	Polyester A + 1.0% PEA from Ex. 2	11.07	2.45	8.77
18	Polyester A + 1.0% PEA from Ex. 3	12.55	1.16	5.11

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Ex #	PE/PEA	Color (b*)	AA genertn (ppm)	
19	Polyester A + 1.0% PEA from Ex. 4	10.08	2.54	6.08
20	Polyester A + 1.0% PEA from Ex. 5	**	3.56	11.14
21	Polyester A + 1.0% PEA from Ex. 6	9.55	1.89	4.35
22	Polyester A + 1.0% PEA from Ex. 7	5.18	2.61	10.61
23	Polyester A + 1.0% PEA from Ex. 8	5.42	1.98	5.58
24	Polyester A + 1.0% PEA from Ex. 10	9.52	0.9	3.23
25	Polyester A + 1.0% PEA from Example 11	10.16	2.91	13.22
26	Polyester A + 1.0% PEA from Example 12	7.16	2.18	12.11
27	Polyester A + 1.0% PEA from Example 13	10.21	1.70	6.07

(1) Polyamide based on m-xylylene diamine and adipic acid

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(2) PEA = polyesteramide

Example 14 shows the AA generated (5.73 ppm and 16.68 ppm) from the control polyester which contained no AA reducing scavengers. Example 15 is PET containing known AA scavengers and displayed reduced AA (1.99 ppm and 8.46). The polyester/polyesteramide of the present invention (Examples 16-27) all display greatly reduced AA which is generally at least as good or better than the polyesters containing the prior art AA scavengers. Moreover, the polyester/polyesteramides of the present invention display improved color as compared to Example 15.

15 Example 28-30 Compounding (PEN) with a polyesteramide.

Poly(ethylene naphthalate) (PEN - 600 g) was dried at 150°C in air and a sample (6.0g) of polyesteramide prepared according to Example 10 was dried at 100°C in a vacuum oven. The samples were physically mixed to give a 1 wt% sample of the polyesteramide in PEN. The mixture was compounded on a Braebender extruder with the main heating zones at 305°C. After compounding the recovered sample was placed into a freezer to minimize loss of acetaldehyde by diffusion until the acetaldehyde could be measured. Acetaldehyde was measured according to the technique described below.

Acetaldehyde concentration after extrusion (Extrusion AA) was determined by grinding the pellets or sheet to 20 mesh or finer and measuring the acetaldehyde concentration by the same gas chromatographic method as described for acetaldehyde generation. The color was measured as described above. The results are shown in Table 2, below.

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TABLE 2:Acetaldehyde Results for  
PEN Compounded Polyesteramides

Ex. #	PE/PEA	color b*	AA extrusn (ppm)
28	PEN control	7.91	16.09
29	PEN + 1% PEA high IV (Ex. 10)	8.05	3.3
30	PEN + 1% PEA low IV (Ex. 10)	8.84	2.39

5  
10 As with the PET, the PEN/polyesteramide  
compositions of the present invention show greatly  
reduced AA (less than one quarter of the PEN control)  
with only very slight increases in the b\* color.

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CLAIMS

1. A composition comprising
- 5           A) 95-99.99 weight percent of a polyester which  
              comprises a dicarboxylic acid component  
              comprising repeat units from at least 85 mol %  
              terephthalic acid, naphthalene-  
              2,6-dicarboxylic acid or mixtures of  
              terephthalic acid and naphthalene-  
10           2,6-dicarboxylic acid and a diol component  
              comprising repeat units from at least 85 mol %  
              ethylene glycol, said mole percents being  
              based on 100 mol % dicarboxylic acid and at  
              least 100 mol % diol, and
- 15           B) 5-0.01 weight percent of a polyesteramide of  
              the formula

$$A(N)_x(D)_y$$

- 20           where A is a dicarboxylic acid selected from  
              aromatic or cycloaliphatic dicarboxylic acids  
              having 8 to 14 carbon atoms or aliphatic  
              dicarboxylic acids having 3 to 24 carbon  
              atoms, N is a diamine having 2 to 24 carbon  
25           atoms, D is a diol having 2 to 14 carbon  
              atoms, X is an integer from 0.01 to 0.99 and Y  
              is an integer from 0.99 to 0.01.

- 30           2. A composition according to Claim 1 comprising  
              98-99.95 weight percent polyester and  
              2-0.05 weight percent polyesteramide.
- 35           3. The composition of claim 1 wherein said  
              dicarboxylic acid component further comprises at  
              least one additional dicarboxylic acid selected

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from the group consisting of aromatic dicarboxylic acids having 8 to 14 carbon atoms, aliphatic dicarboxylic acids having 4 to 12 carbon atoms and cycloaliphatic dicarboxylic acids having 8 to 12 carbon atoms.

4. The composition of claim 3 wherein said at least one additional dicarboxylic acid is selected from the group consisting of phthalic acid, isophthalic acid, naphthalene-2,6-dicarboxylic acid, cyclohexanedicarboxylic acid, cyclohexanediacetic acid, diphenyl-4,4'-dicarboxylic acid, succinic acid, glutaric acid, adipic acid, azelaic acid, sebacic acid, and the like.
5. The composition of claim 1 wherein said diol component further comprises at least one additional diol selected from the group consisting of cycloaliphatic diols having 6 to 20 carbon atoms or aliphatic diols having 3 to 20 carbon atoms.
6. The composition of claim 7 wherein said at least one additional diol is selected from the group consisting of diethylene glycol, triethylene glycol, 1,4-cyclohexanedimethanol, propane-1,3-diol, butane-1,2-diol, pentane-1,5-diol, hexane-1,6-diol, 3-methylpentanediol-(2,4), 2-methylpentanediol-(1,4), 2,2,4-trimethylpentane-diol-(1,3), 2-ethylhexanediol-(1,3), 2,2-diethylpropane-diol-(1,3), hexanediol-(1,3), 1,4-di-(hydroxyethoxy)-benzene, 2,2-bis-(4-hydroxy-cyclohexyl)-propane, 2,4-dihydroxy-1,1,3,3-tetramethyl- cyclobutane,

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2,2-bis-(3-hydroxyethoxyphenyl)-propane, and  
2,2-bis-(4-hydroxypropoxyphenyl)-propane.

7. The composition of claim 1 wherein said  
5 dicarboxylic acid of said polyesteramide is  
selected from the group consisting of terephthalic  
acid, isophthalic acid, 1,4-cyclohexanedicarboxylic  
acid, 1,3-cyclohexane-dicarboxylic acid,  
2,6-naphthalene dicarboxylic acid, succinic acid,  
10 adipic acid, azelaic acid, sebacic acid and  
mixtures thereof, and/or said diol of said  
polyesteramide is selected from the group  
consisting of ethylene glycol, diethylene glycol,  
triethylene glycol, propane diol,  
15 1,4-cyclohexanediol, 1,4-butanediol,  
2,2-bis-(4-hydroxy-cyclohexane)-propane,  
1,3-(2,2-dimethyl)propane diol,  
1,4-cyclohexanedimethanol, 1,4-di-(hydroxyethoxy)  
benzene, hydroquinone, bisphenol A and mixtures  
20 thereof.
8. The composition of claim 1 wherein said diamine is  
selected from the group consisting of ethylene  
diamine, propane diamine, butane diamine, hexane  
25 diamine, 1,4-cyclohexanediamine, 1,3-cyclohexane  
diamine, 1,4-phenylene diamine, methylene  
dianiline, 1,3-xylylene diamine,  
1,3-(2,2-dimethyl)propane diamine, 1,4 and  
1,3-cyclohexanebismethylamine, 1,4-xylylene  
30 diamine, bis(p-aminocyclohexyl)methane and mixtures  
thereof.
9. The composition according to Claim 1 wherein said  
polyester has an I.V. of 0.5-1.0; said  
35 polyesteramide has a number average molecular

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weight of 1000–100,000, and wherein X is an integer from 0.25 to 0.74 and Y is an integer from 0.74 to 0.25.

- 5      10. An article selected from a molded article, beverage bottle, film or sheet comprising the composition of Claim 1.

# INTERNATIONAL SEARCH REPORT

Intern: 1 Application No

PCT/US 97/01638

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C08L67/02 //(C08L67/02,77:12)

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	RESEARCH DISCLOSURE, vol. 283, no. 006, 10 November 1987, HAVANT GB, pages 633-636, XP000027721 ANONYMOUSLY: "Polyester/Polyesteramide Blends" see page 633, left-hand column, line 1 - right-hand column, line 42	1-10
X	EP 0 143 953 A (THE DOW CHEMICAL COMPANY) 12 June 1985 see claims 1-10	1-10

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

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"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

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Date of the actual completion of the international search

21 May 1997

Date of mailing of the international search report

6. 06. 97

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## INTERNATIONAL SEARCH REPORT

Information on patent family members

Internal	Application No
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PCT/US 97/01638

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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		CA 1262390 A	17-10-89
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